



Overview of Ozone/VOC/NO_x Production and Transport in Northern/Central California

Shaheen R. Tonse

EET Division

Lawrence Berkeley National Laboratory

Workshop on Passive Catalytic Oxidation of Air Pollution

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Overview



- **Ozone: Sources, Sinks**
- **Two slides about modeling ozone**
- **Ozone chemistry - simplified picture**
- **Emissions of Ozone Precursors**
- **Sensitivity to Precursor Emissions**
- **Points of Relevance to Catalytic Oxidation**

Ozone: Sources, Sinks (1)



- Secondary pollutant by photochemical reactions:



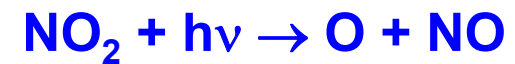
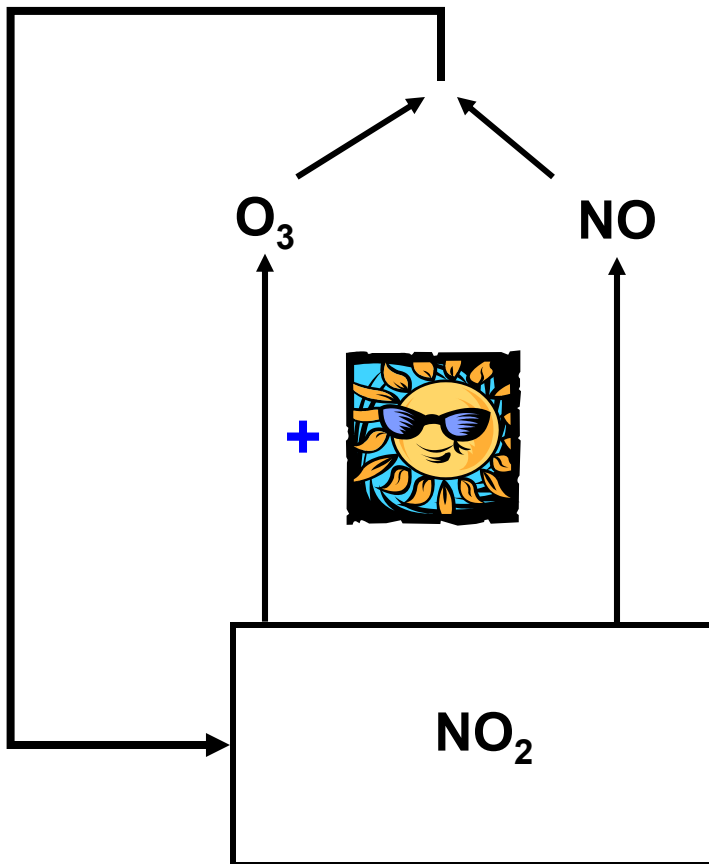
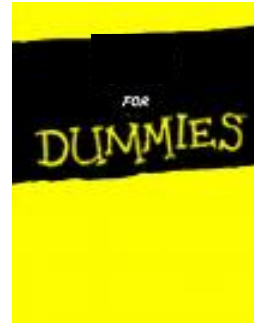
- Concentrations on a good day ~20-40 ppb
- A bad day: 120-150ppb
(<http://www.arb.ca.gov/aqd/aqinfo.htm>)
- US EPA limit: Peak ozone \geq 120ppb in any given hour or ozone \geq 80ppb averaged in any 8-hour window

Ozone: Sources, Sinks (2)



- Human health
- Agricultural health
- Lifetime in atmosphere about 1 week, destroyed by photolysis:
 - $O_3 + h\nu \rightarrow O_1D$ followed by
 - $O_1d + H_2O \rightarrow 2 OH$
- Also dry deposition to surface

O₃ Production



followed by

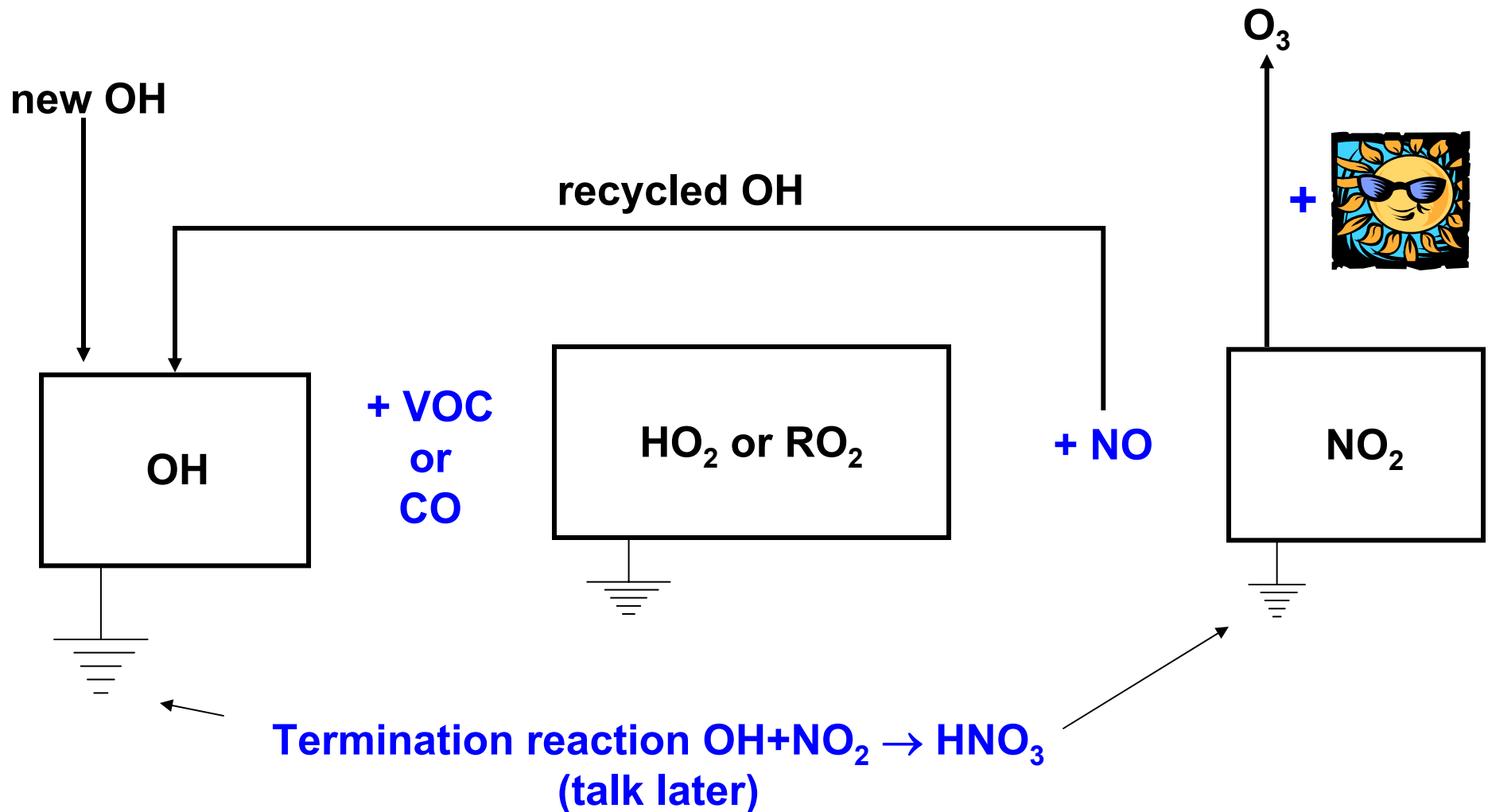


and



.....should not see more
than about 50ppb if this
is all there is.

O₃ Production



VOCs are key to conversion of NO to NO₂ without destruction of an O₃ molecule in the process

Modeling Approach



CMAQ Model: 3D

advective/diffusive
transport and photo-
chemistry code

SAPRC99 gas phase
mechanism (72 lumped
and individual species,
250 reactions)

Cells 4km to a side
96 cells east-west
117 cells north-south
27 vertical layers from up to
16km

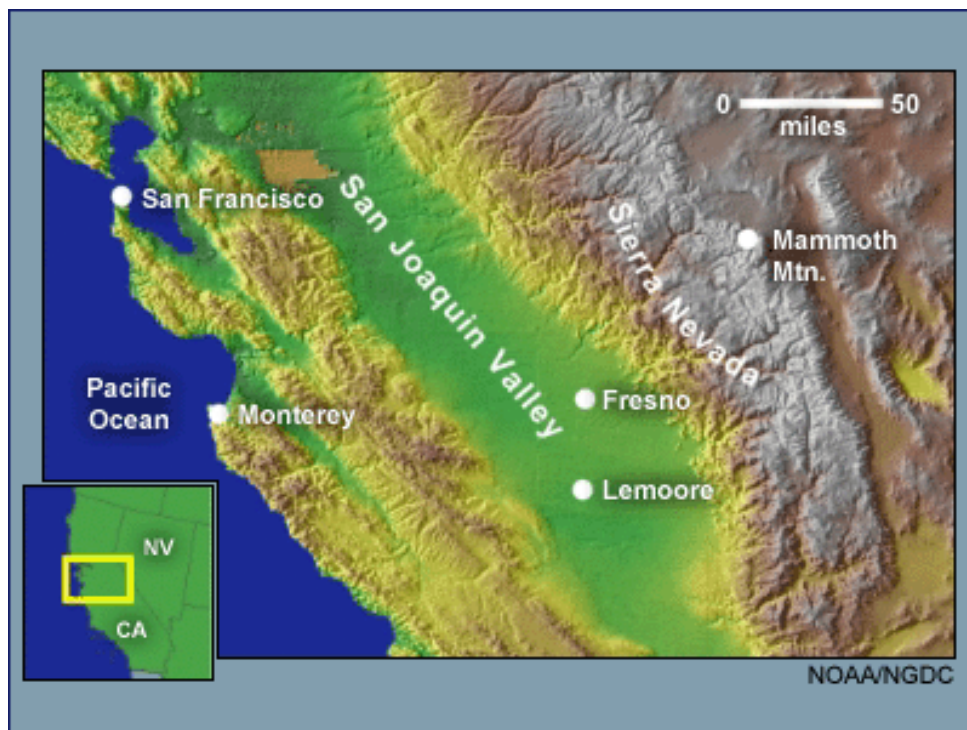
**Meteorological
Model**

**Emissions
Model**

**Photochemical
Air Quality Model**

**Analysis of
Model Outputs**

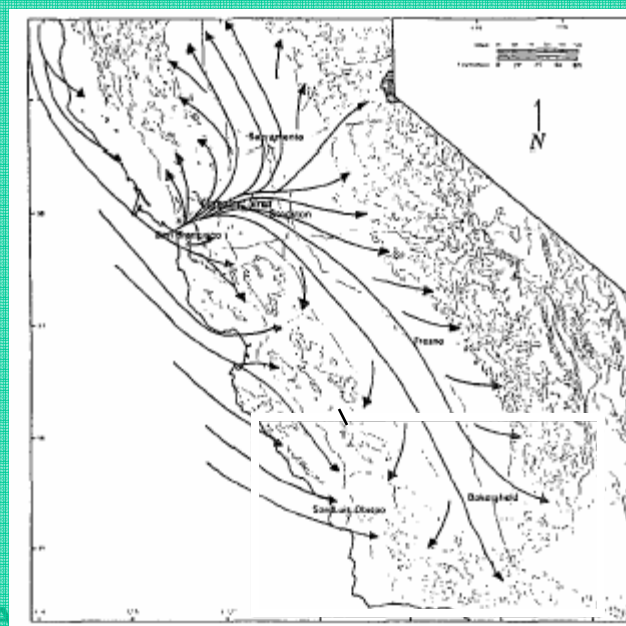
Geography of the Study Area



The SJV is surrounded by the Diablo and Coast Ranges on the west and the Sierra Nevada foothills on the east.

The mouth of the San Francisco Bay forms a gap in the coastal range that allow wind to blow pollutants into the valley from other regions.

The predominant summer wind flow pattern



Niccum *et al.* 1995

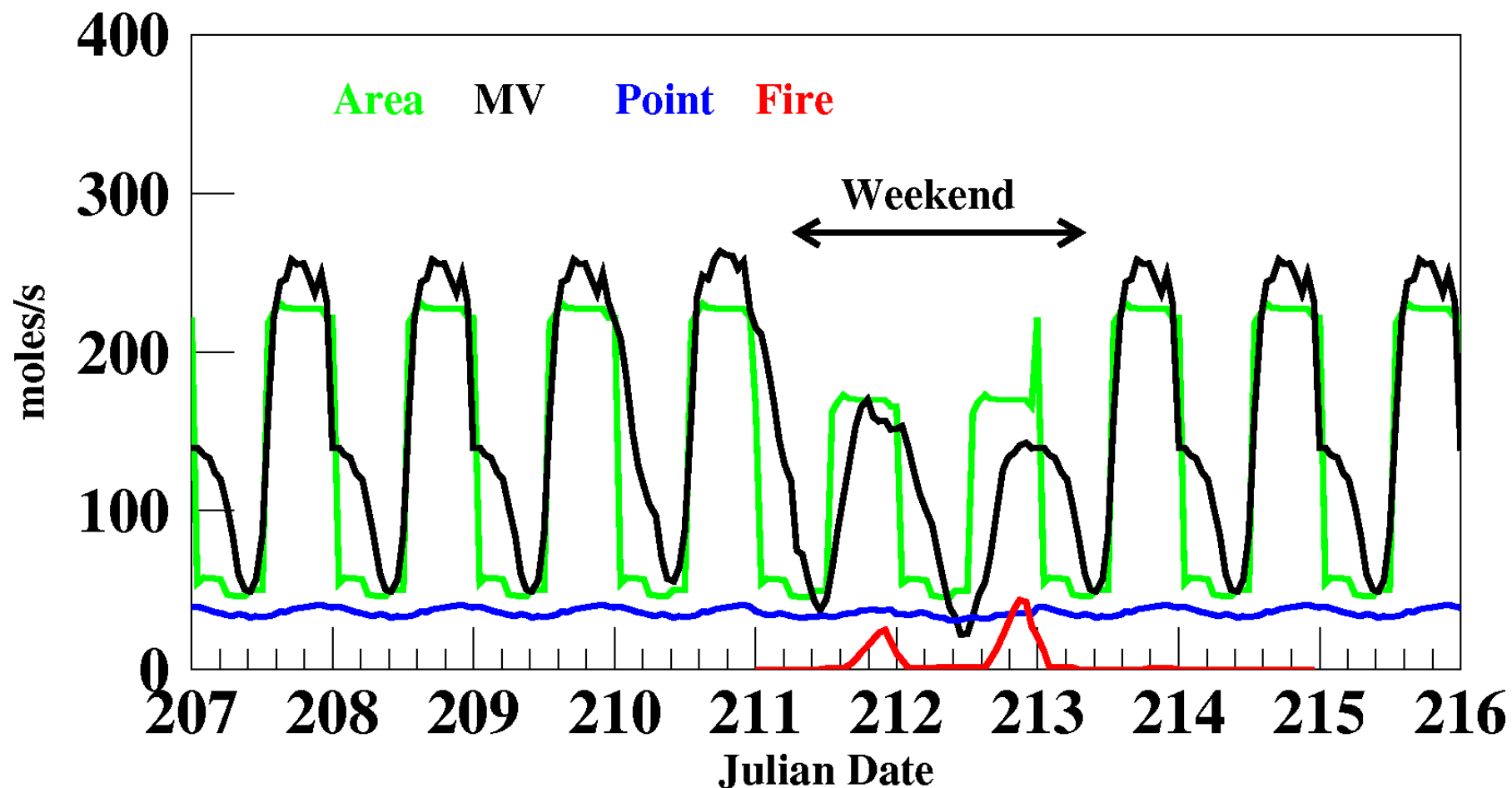
Emission Inventory



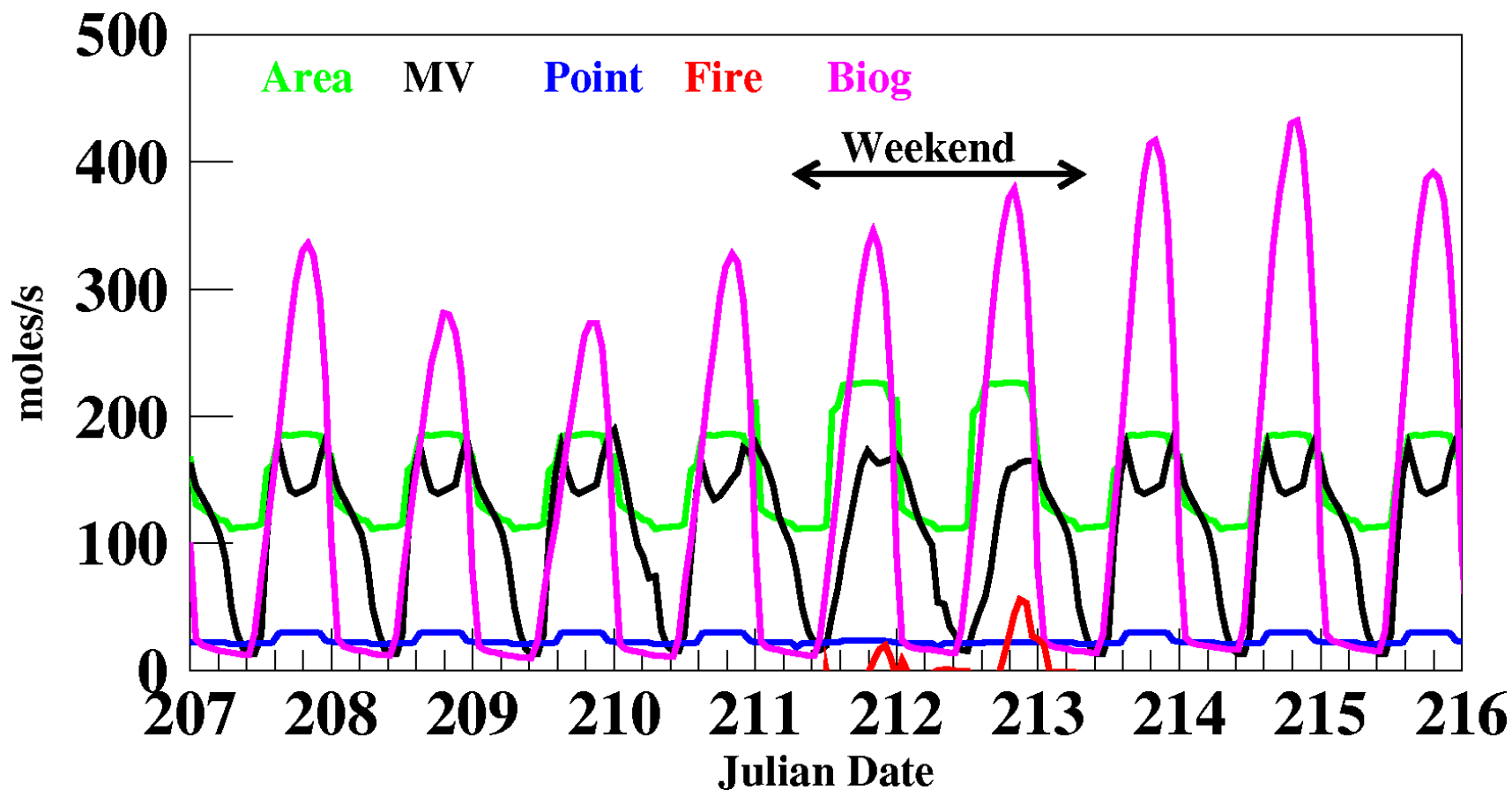
- > **Mobile** (automobiles, diesel trucks)
- > **Point** (cement, power plants, industry)
- > **Area** (residential, off road vehicles, business)
- > **Biogenic** (trees, agriculture)
- > **Forest Fires**

NOx emissions by category

(next 3 figures from Phase 1 report)



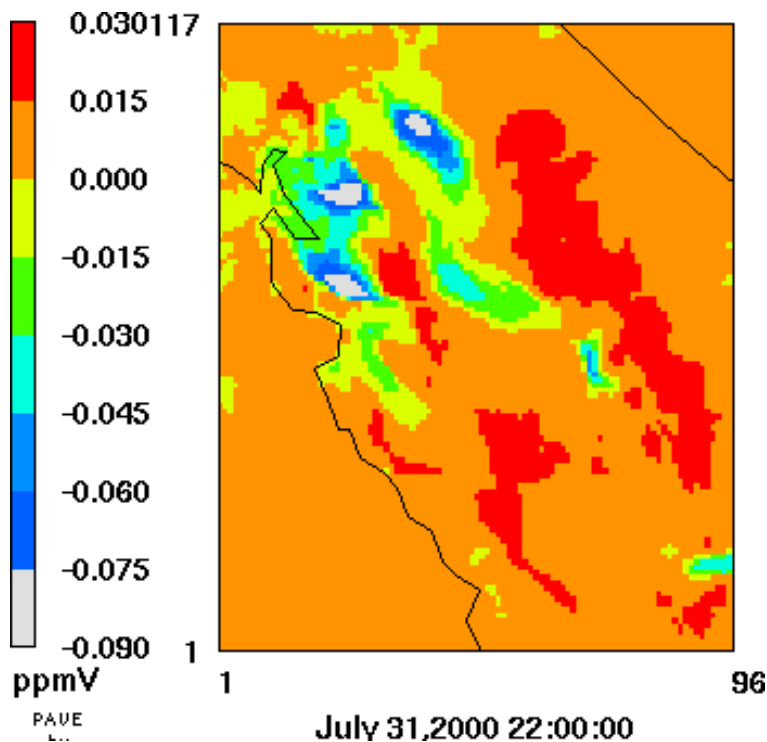
VOC emissions by category



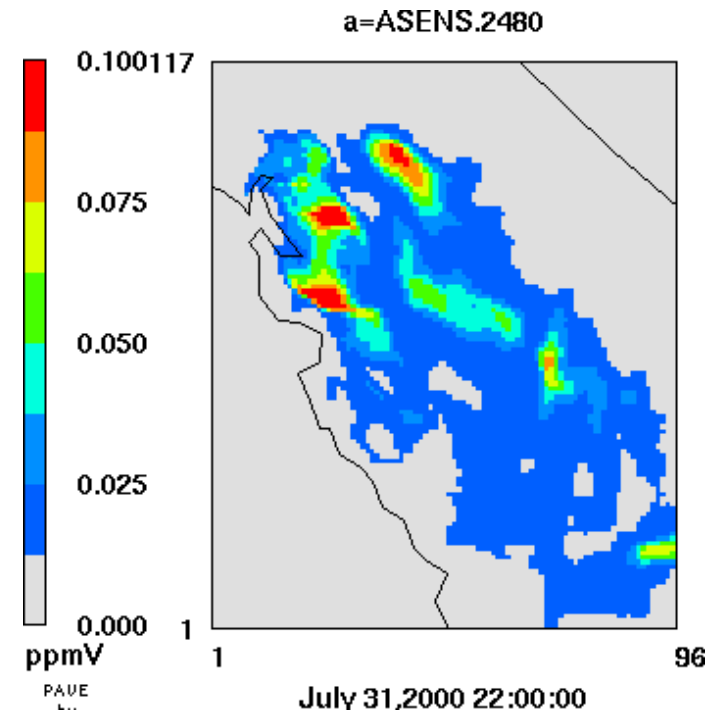
How does Ozone change w.r.t. Precursors: Sensitivity Analysis with CMAQ Decoupled Direct Method

Sensitivities w.r.t. emissions

$dO_3/d(NO_x)$: Sometimes
negative, sometimes positive.



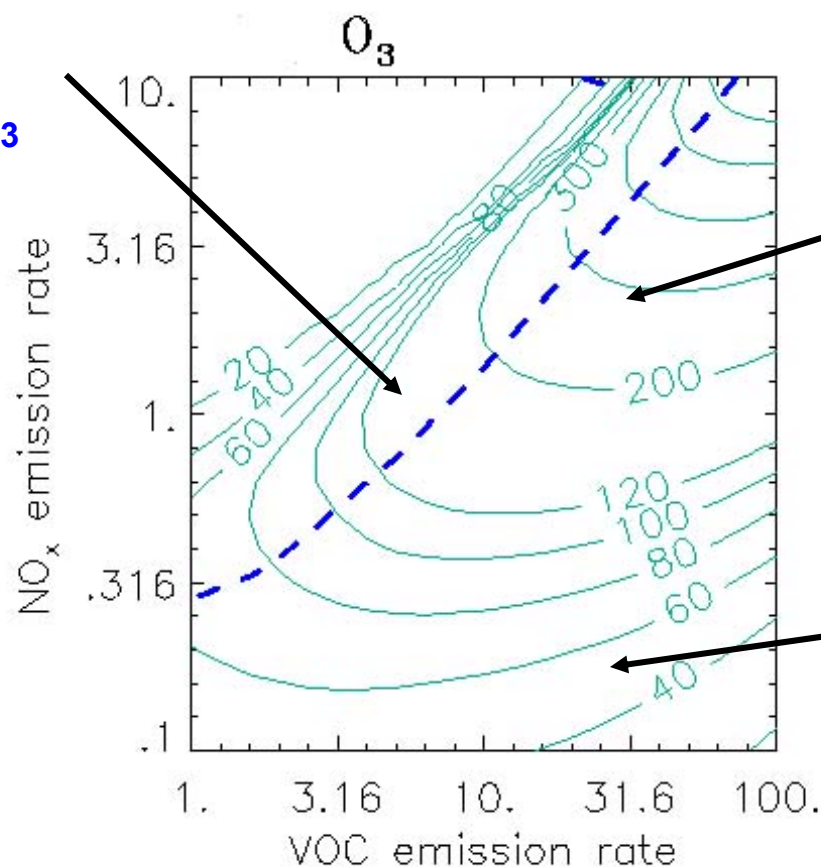
$dO_3/d(VOC)$: Always
positive



Ozone Isopleth Plot (Googled)



Urban area
 $\text{OH} + \text{NO}_2 \rightarrow \text{HNO}_3$



Houston, TX:
autos +
petrochemical
industry

Forested area

Points of Relevance to Catalytic Oxidation



Main benefit of reducing VOC and NO_x concentrations is to reduce ozone. (California does not violate air quality standards for VOC and NO_x)

Most NO_x emissions are in the form of NO. NO_2 is 5-10%

NO oxidation: If only to NO_2 then there will be an increase in ozone

Careful about reducing NO_x if an increase in ozone will result

VOC oxidation: During ozone production a primary VOC is successively oxidized in steps, producing intermediate compounds (aldehydes, ketones) in the process